

EFFECT OF AERATION RATES ON SIMULTANEOUS NITRIFICATION AND DENITRIFICATION IN INTERMITTENT AERATED BIOREACTOR

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Abstract– Eutrophication caused by high nitrogen wastewater has become a serious problem in many countries. One of the current common applied technology for nitrogen removal is biological process. In this work, an intermittent aerated bioreactor was developed to treat high nitrogen wastewater. The effect of different aeration rates on the reactor performance was also investigated. The results showed that at the low aeration of 0.5 L/min, the nitrogen removal efficiency reached the peak of 65%, while the nitrification and denitrification rates were approximately 0.2 and 1.8 mg/L-min, respectively. Despite of the higher nitrification rates at higher aeration rates, the denitrification rates were found to decrease because of competitive microorganisms growth and the lack of carbon source. This resulted in the drop of the overall nitrogen efficiency with the increasing aeration rates.

INTRODUCTION

Nitrogen and phosphorus are the essential nutrients for living things. However, when these nutrients are present at high concentrations in water bodies, they can cause the explosive growth of aquatic plants, typically algae, which leads to the oxygen depletion in water and the death of aquatic organisms as a result. This phenomenon is usually known as algae bloom or eutrophication. One of the main sources of nutrients discharge which results in eutrophication is the domestic wastewater (Gong *et al.*, 2012; Wang *et al.*, 2014). In the typical domestic wastewater, the concentration of nitrogen can be up to 44 mg/L of NH₄-N (Nguyen *et al.*, 2014a; Nguyen *et al.*, 2014b). Meanwhile, less than 0.5 mg/L of NH₄-N should be found in surface water (PCD, 2014). Eutrophication is a big and challenging problem in many countries all over the world. One of the reasons is the lack of a both highly effective and economical technology to remove nitrogen from the wastewater.

Among the most common used techniques these days, biological process has been considered to be able to obtain good performance and relatively cost

effectiveness (Huang *et al.*, 2013; Yao *et al.*, 2013). Liu *et al.* (2013) revealed that the biological nitrogen removal can be enhanced by intermittent aeration mode, and the nitrification microbial process plays an important role. The growth of ammonia-oxidizing bacteria and nitrite oxidizing bacteria which are nitrifying microorganisms was reliable to increase under the intermittent aeration (Fan *et al.*, 2013). However, the different intermittent periods can cause a shift of diverse microbial community (such as phylum *Proteobacteria*, *Firmicutes* and *Bacteroidetes*) in the nitrogen removal system (Guadie *et al.*, 2014). Therefore, the aim of this work is to investigate the nitrogen removal efficiency of an intermittent aerated bioreactor under a typical intermittent period. The effects of different air supply rates on the bioreactor performance and microbial community are also examined.

METHODOLOGY

Synthetic Wastewater

In order to minimize variability in the

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experiment, the synthetic wastewater was used in this study by mixing the following chemicals (g/L); NH_4Cl 0.15 g, NaHCO_3 0.48 g, KH_2PO_3 0.02 g, MgSO_4 0.06 g, CaCl_2 0.36 g, FeSO_4 0.003 and trace element 0.5 mL (Guo *et al.*, 2013). The concentration of $\text{NH}_4\text{-N}$ was approximately 40 mg/L. In the meantime, other nitrogen forms, particularly $\text{NO}_2\text{-N}$ and $\text{NO}_3\text{-N}$ have the concentration of less than 1 mg/L.

Bioreactor Set-Up and Operation

Active sludge was obtained from the wastewater treatment plant of Wongtong Hospital, Phitsanulok, Thailand and acclimatised in the synthetic wastewater for a month. A bioreactor with 24 cm in diameter and 40 cm in height was set up using 2 L of acclimatized sludge and 8 L of synthetic wastewater. During operation, the sludge was not disposed (SRT = 0), and the MLSS and MLVSS were approximately 4,600 and 3,200 mg/L. A cycle of intermittent aeration supply was 24 hours. During the aeration period, air was supplied for two hours at three different flow rates; low aeration (0.5 L/min), medium aeration (1.0 L/min) and high aeration (2.0 L/min). During the non-aeration period, no air was supplied to the bioreactor for two hours, thus the dissolved oxygen (DO) concentration was sharply dropped to 0.5 mg/L. The acetate solution was added in the first non-aeration period as a carbon source for the denitrification process. To minimize the operation cost, a low C/N ratio of 1.5 was controlled (Khanitchaidecha *et al.*, 2010). After that the aeration and non-aeration periods of two hours were continued to finish 24 hours (Fig. 1). The nitrogen removal efficiency of the bioreactor was calculated using Eq. 1.

$$\text{Efficiency} = \left(1 - \frac{[\text{NH}_4\text{-N}]_{\text{eff}} + [\text{NO}_3\text{-N}]_{\text{eff}} + [\text{NO}_2\text{-N}]_{\text{eff}}}{[\text{NH}_4\text{-N}]_{\text{inf}}}\right) \times 100 \quad \dots (1)$$

where, $[\text{NH}_4\text{-N}]_{\text{inf}}$ = concentration of ammonium in the synthetic wastewater

$[\text{NH}_4\text{-N}]_{\text{eff}}, [\text{NO}_3\text{-N}]_{\text{eff}}, [\text{NO}_2\text{-N}]_{\text{eff}}$ = concentrations of ammonium, nitrate and nitrite in the treated water

After one cycle of treatment (24 hours) finished, the treated wastewater in the bioreactor was drained out and replaced by the new synthetic wastewater with the same constituents and volume. The bioreactor was operated for three months and the average efficiency of nitrogen removal was achieved.

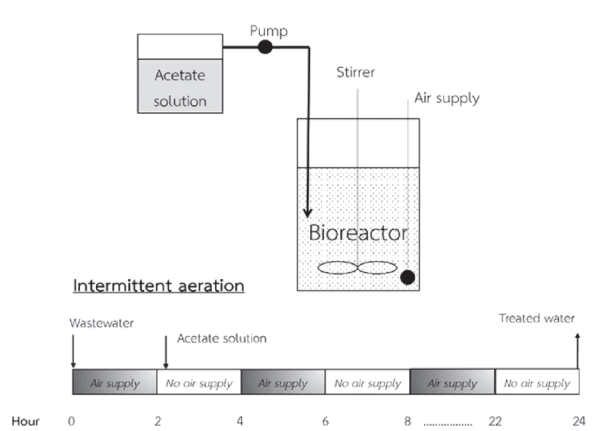


Fig. 1. Schematic diagram of bioreactor

Nitrification and Denitrification Rates Determination

For the nitrification rate determination, the $\text{NH}_4\text{-N}$ wastewater of 40 mg/L was added in the bioreactor which continuously stirring and aerating. The water was sampled every 10 min and analysed for changing $\text{NH}_4\text{-N}$, $\text{NO}_2\text{-N}$ and $\text{NO}_3\text{-N}$. The rate of $\text{NH}_4\text{-N}$ reduction can refer to nitrification rate. On the other hand, the denitrification rate was determined by adding the $\text{NO}_3\text{-N}$ wastewater of 40 mg/L into the bioreactor. The bioreactor was continuously stirring and no aerating. The acetate solution was added in the beginning to maintain the C/N ratio of 1.5. Then the water was sampled every 10 min and analysed for changing $\text{NO}_2\text{-N}$ and $\text{NO}_3\text{-N}$.

RESULTS AND DISCUSSION

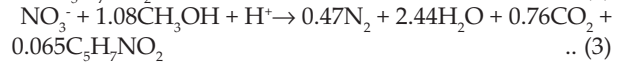
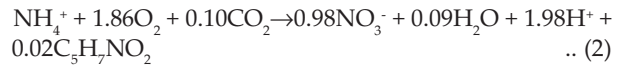
Performance of Intermittent Aerated Bioreactor and Nitrogen Removal Mechanisms

The bioreactor was operated under intermittent aeration at various aeration rates of 0.5 (low), 1.0 (medium) and 2.0 (high) L/min, respectively. At the low aeration rate, the $\text{NH}_4\text{-N}$ and $\text{NO}_2\text{-N}$ was not found in the treated water, however the $\text{NO}_3\text{-N}$ ranging of 10-20 mg/L was occurred (data not shown). The average efficiency was around 65% (Fig. 2). The water samples were taken to determine the mechanisms occurred in the bioreactor for nitrogen removal, and the results are shown in Figure 3a. In the first aeration period (hour 0-2), the $\text{NH}_4\text{-N}$ was sharply decreased from 40 to 27 mg/L,

while the $\text{NO}_3\text{-N}$ was immediately increased to 16 mg/L. This indicated the occurrence of nitrification process and the sufficient oxygen supply to convert $\text{NH}_4\text{-N}$ to $\text{NO}_3\text{-N}$ with no intermediate form of $\text{NO}_2\text{-N}$. From Figure 3b and 3c, the reductions of DO and pH refer that large volume of oxygen was consumed for nitrification and H^+ was generated. The theoretical nitrification equation was present in Eq. 2 (Khanitchaidecha *et al.*, 2013).

In the first non-aeration (hour 2-4), the $\text{NH}_4\text{-N}$ continuous to decrease to 18 mg/L. Although air supply was stopped, however the remaining oxygen can oxidise some $\text{NH}_4\text{-N}$ to $\text{NO}_3\text{-N}$. Due to the low oxygen and existence of organic carbon, the occurred $\text{NO}_3\text{-N}$ was simultaneously removed by denitrification process, as supported by increasing pH from generated OH. The theoretical nitrification equation was present in Eq. 3 (Khanitchaidecha *et al.*, 2011). The results reveal that the nitrification and denitrification was occurred simultaneously in this period. The $\text{NH}_4\text{-N}$ was continuously oxidised in the second aeration (hour 4-6), until its concentration became zero. However, due to insufficient organic carbon (~5 mg COD /L remained), the $\text{NO}_3\text{-N}$ still remained in the treated

water (data not shown).



Performance at Various Aeration Rates

Figure 2 presents the average efficiency of nitrogen removal during three months of operation under different aeration rates. It can be seen that the highest efficiency was obtained at the low aeration rate of 0.5 L/min. The nitrogen removal efficiency decreased from 65% to 47% and 38% with the increase of aeration rates from 0.5 L/min to 1.0 and 2.0 L/min, respectively. These results have proved that the aeration rate affected to the nitrogen removal efficiency of the intermittent aerated bioreactor. The observed decrease of nitrogen removal efficiency can be explained based on the results of $\text{NH}_4\text{-N}$, $\text{NO}_2\text{-N}$ and $\text{NO}_3\text{-N}$ concentration analysis (Table 1). Although the nitrogen removal efficiency in this study was fairly low, the lower air supply and carbon addition than other studies were significant reasons (as summarised in Table 2). In addition, the use of supporting media such as polyurethane foam for increasing microorganisms and enhancing reactor performance was found in the previous studies.

The nitrogen removal mechanisms occurred in this bioreactor can be divided into two stages that are nitrification and denitrification. Nitrification is the process in which $\text{NH}_4\text{-N}$ is transferred into $\text{NO}_2\text{-N}$ and then $\text{NO}_3\text{-N}$ in the presence of oxygen. Meanwhile, in the denitrification process, in the presence of carbon source and in the absence of oxygen, $\text{NO}_3\text{-N}$ is transferred into $\text{NO}_2\text{-N}$ and then nitrogen gas which is finally released into the air. Table 1 shows the removal rates of $\text{NH}_4\text{-N}$, $\text{NO}_2\text{-N}$ and $\text{NO}_3\text{-N}$ as the indicators of nitrification and denitrification processes at different aeration rates.

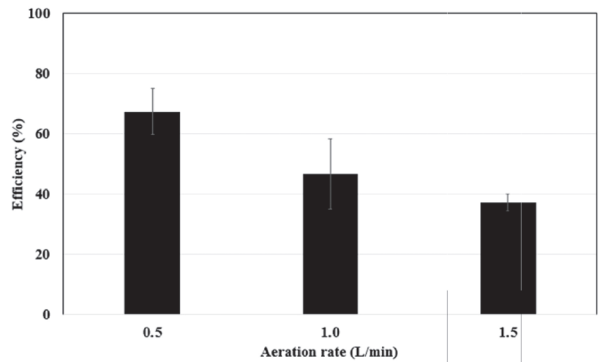


Fig. 2. Average efficiency of intermittent aerated bioreactor at various aeration rates.

Table 2. Comparison of the reactor performance and operating condition of this study and previous studies

Intermittent hours (on/off)	Influent $\text{NH}_4\text{-N}$ (mg/L)	C/N ratio	Air supply	Efficiency (%) (L/min)	Reference
0.75/0.25	40-60	1.5	8	82	Guadieet <i>al.</i> , 2014
1/1	50	1.0	N/A*	57	Lim <i>et al.</i> , 2012
1/2	25	1.2	4	80	Cho <i>et al.</i> , 2007
1/5	40	1.7	2	80	Fan <i>et al.</i> , 2013
2/1	25	3.2	1.4	82	Mouraet <i>al.</i> , 2012
2/2	40	1.5	0.5	65	This study

Remark: * DO 6-7 mg/L

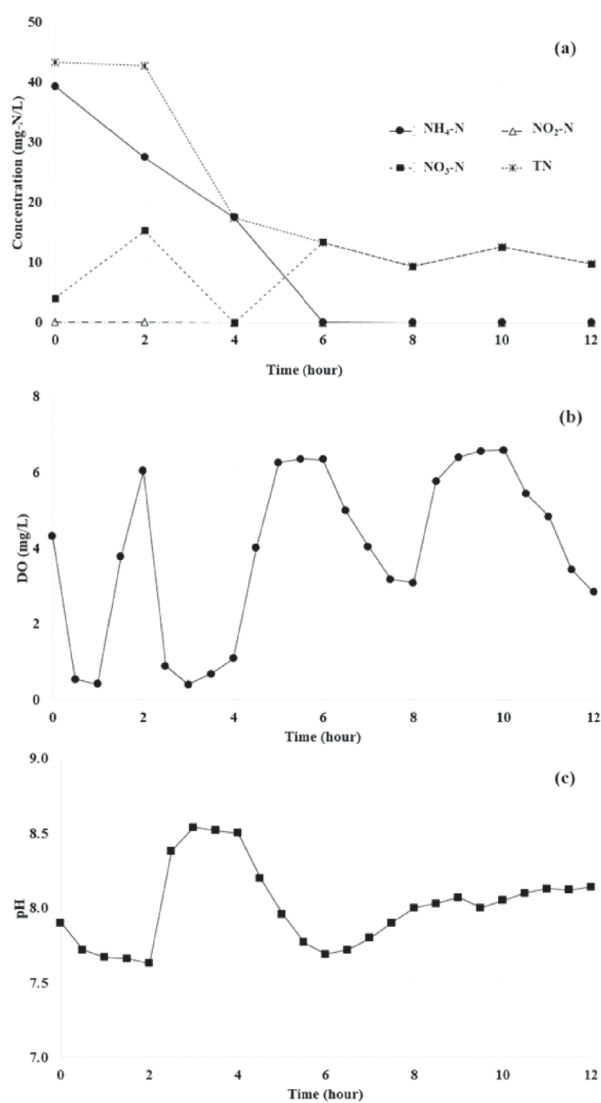


Fig. 3. Profiles of (a) nitrogen, (b) DO and (c) pH at low aeration rate.

It can be seen that the nitrification rate increased with the increasing aeration rates. This is because the activity of the microorganism driving nitrification process was enhanced under high oxygen conditions (Jia *et al.*, 2013). In the next stage of denitrification, although there was no aeration, the excess of oxygen was still remained in the bioreactor. This amount of oxygen not only inhibited the activity of denitrification microorganisms but also created conditions for the growth of competitive microorganisms, typically heterotroph. Heterotroph consume both oxygen and carbon source, which led to the shortage of carbon for denitrification microorganisms. Therefore,

denitrification process was inhibited, leading to the drop of denitrification rates. This explanation is also supported by the change of the sludge colour from dark brown to light brown during the operation process, which refers to the change in dominant microorganism's community in the bioreactor. The SVI became to increase from 54 mL/g at the low aeration to 72 mL/g at the high aeration.

Furthermore, it has to be noted that in addition to the growth of heterotroph, nitrogen could not be completely removed from the wastewater also because of the insufficient carbon source. The bioreactor fed on acetate solution as carbon source only one time when the bioreactor was set up. This carbon source was sufficient for the first non-aeration period (hour 2-4). However, in the next periods, acetate was consumed and became insufficient for the denitrification process to occur effectively. Thus the NO₃-N still remained while the NH₄-N was completely removed after treatment.

CONCLUSION

In the present work, the intermittently aerated bioreactor for nitrogen removal has successfully developed. The results showed that the set-up bioreactor was able to remove 65% of nitrogen at the aeration rate of 0.5 L/min. However, the efficiency decreased with the increasing aeration rates due to the growth of competitive microorganisms. Additionally, the lack of carbon source was found to be one of reasons of the incomplete nitrogen removal. In conclusion, this work has proved that the nitrogen removal process consisted of two stages, nitrification and denitrification, which were affected by different aeration rates.

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